

Effects of Electron Donor Additives to Sewage Sludge on Biomethanation of Gaseous Carbon Dioxide

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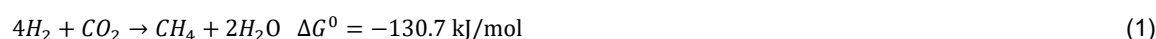
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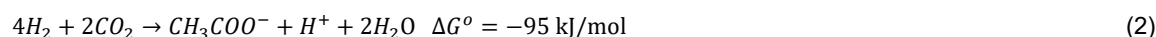
This study investigated the effect of concentration on biomethanation in electron donor additives (zero valent iron (ZVI) 10-40 g/L, activated carbon (AC) 1.0- 4.0 g/L and Fe₃O₄ 0.25-1.0 g/L) when carbon dioxide (CO₂) is the only carbon source in anaerobic digestion., and the researchers estimated the pathway of CO₂. ZVI had 43-49 % biomethanation of injected CO₂ but was not observed in AC and Fe₃O₄. It has been confirmed that the rate of hydrogen (H₂) production from ZVI depending on its size. Furthermore, when there is a shortage of CO₂, inhibition occurs due to the hydrogen generated from ZVI. The reason for the inhibition of methane (CH₄) production in AC and Fe₃O₄ was found to be due to volatile fatty acids (VFAs) accumulation. The use of additives induces the conversion of CO₂ to VFAs and inorganic carbon, which are necessary for biomethanation. The addition of ZVI increased the growth of hydrogenotrophic methanogens, thereby enabling CH₄ production without VFAs accumulation.

1. Introduction

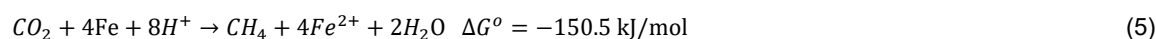
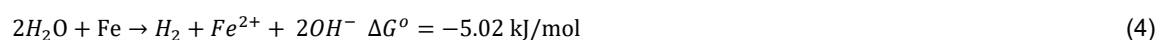
Anaerobic digestion is used to stabilize a variety of organic wastes, including sewage sludge, animal manure, and food waste (Dong et al., 2019). Biogas from anaerobic digestion is considered a renewable energy alternative to natural gas, and biogas use may contribute to reduced greenhouse emissions (Dong et al., 2022). Raw biogas (40-70 % methane (CH₄) and 30-60 % carbon dioxide (CO₂)) must be upgraded to biomethane (>95 % CH₄) for practical use due to lower CH₄ content (Andronikou et al., 2022). Numerous recent studies have investigated the method for upgrading biogas based on hydrogen (H₂) and CO₂ usage by hydrogenotrophic methanogens by Eq(1) (Angelidaki et al., 2018).



This biochemical conversion of CO₂ is also carried out by homoacetogens and acetoclastic methanogens (Dong et al., 2022). Homoacetogen can use CO₂ and H₂ for acetic acid production (Eq(2)) and acetoclastic methanogens can utilize acetic acid for the generation of CO₂ and CH₄ (Eq(3)) (Vyrides et al., 2018).



One of the typical biological methanation technologies is the use of H₂ or electron donors, and many studies have been conducted to generate H₂ and electron donors using additives (Dong et al., 2019). Zero valent Iron (ZVI, Fe⁰) was mainly used in the study to generate H₂ and electron donors. ZVI has improved anaerobic digestion with H₂ generation (Eq(4)) and direct interspecies electron transfer (DIET) (Eq(5)) (Dong et al., 2022).



Activated carbon (AC) and Fe_3O_4 were mainly used as electron donors. AC is an inexpensive material that can significantly increase CH_4 production by accelerating DIET and volatile fatty acids (VFAs) consumption (Li et al., 2019). Fe_3O_4 has the ideal potential to accelerate CH_4 production through the decomposition of complex organic matter. Fe_3O_4 can promote CH_4 production through DIET (Zhao et al., 2018).

Many studies have been conducted to improve the efficiency of anaerobic digestion using additives (Wang et al., 2016). Existing research has predominantly focused on materials that generate H_2 , and there is a lack of research on electron donors in the context of CO_2 conversion to CH_4 using a single carbon source (Andronikou et al., 2022). The study focused on investigating the role of electron donors using AC, Fe_3O_4 , and ZVI, as electron donors for the biochemical conversion of CO_2 to CH_4 in anaerobic digestion experiments. This study aimed to understand how the electron donors could influence the biochemical process of converting CO_2 to CH_4 and to determine the impact of electron donors on the pathways of CO_2 in biomethanation.

2. Materials and Method

2.1 Anaerobic biomethanation batch test

125 mL serum bottles with a working volume of 50 mL were used for batch lab-scale anaerobic biomethanation experiments. The working volume contains 20 mL of nutrient mineral medium, 20 mL of anaerobic inoculum, and 10 mL of HEPES (4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid) buffer solution. According to a prior study, nutrient mineral media for methanogens was created (Dong et al., 2019). HEPES buffers were used to maintain neutral pH (Dong et al., 2019). In a 1 L batch reactor operating at 37 °C, pH 7.1-7.3, and 150 rpm, anaerobic inoculum was cultivated with anaerobic digested sludge for 2 weeks to remove biodegradable organic matter from the sludge. In every day, generated biogas was removed and the pH was adjusted and purged with nitrogen gas (99.99 %) for 1 min. The concentration of volatile suspended solids (VSS) in the anaerobic inoculum was $5,083 \pm 144$ mg/L and the initial pH of the liquid including nutrient mineral medium, anaerobic inoculum, and HEPES buffer solution was 7.12 ± 0.01 , before running the experiment. After pH measurement, AC (0.2-1 mm, Fujifilm wako pure chemical) concentration of 1, 2, and 4 g/L, Fe_3O_4 (50-100 nm, Sigma Aldrich) concentration of 0.25, 0.5, and 1 g/L, ZVI (<150 μm , Fujifilm wako pure chemical) concentration of 10, 20 and 40 g/L were added. Without adding materials, the control was prepared. After being sealed with silicon septa, all serum bottles underwent a one-minute nitrogen gas (99.99 %) purge. The samples, which contained nutritional mineral medium, anaerobic inoculum, HEPES buffer solution, and additives, were injected with 20 mL of CO_2 gas (99.99 %), and a push-pull method was used to extract the same amount of headspace gas. All serum bottles were placed in a water bath (BT300, Yamato scientific., Japan) at 37 °C and shaken at 140 rpm. Triplicates of each anaerobic batch experiment were carried out. During the experiment, the gas composition was analyzed, and the gas volume was measured using a syringe. After the experiment, the samples were analyzed for pH, total inorganic carbon (TIC), VFAs, and microbial community structure.

2.2 Analytical techniques

A pH meter (HM-31P, DKK-TOA Co., Japan) was used to measure the pH. VSS was measured according to the standard method from American public health association (American public health association, 1998)

The concentrations of CH_4 , CO_2 , and H_2 of gas samples were analyzed using a Gas Chromatograph (490 Micro GC, Agilent Technologies, USA) equipped with a Thermal Conductivity Detector (TCD). PoraPLOT Q Columns was used with helium as the carrier gas at 40 °C and 190 kPa to detect CO_2 . Molsieve 5Å columns was used with argon as the carrier gas at 80 °C and 170 kPa to detect CH_4 and H_2 .

Total inorganic carbon (TIC) analysis was conducted by a TOC analyzer (TOC-V_{CSH}, Shimadzu Corp., Japan). 680 °C combustion catalyst oxidation and non-dispersive infra-red (NDIR) detectors method was adopted.

The concentrations of VFAs (Lactic, Formic, Acetic, Propionic, Isobutyric, Isovaleric, Valeric, and Butyric) was analyzed by high-performance liquid chromatography (L-20AD, Shimadzu, Japan) equipped with a CDD-10AVP conductivity detector, a Shimadzu SIL-20AC autosampler, and a CTO-10AC VP column oven. Shim-pack SCR-102H column (300 mm × 8.0 mm × 7 μm) was used with 5 mM perchloric acid rate of 1.0 mL/min at 40 °C. The injection volume was 20 μL . The samples were centrifuged at $3,500 \times g$ for 10 min and filtered with 0.45 μm syringe filters.

2.3 Microbial community structure analysis

For the microbial community structure analysis, the samples that were collected at the finish of the experiment are analyzed using polymerase chain reaction (PCR) analysis. DNA extraction was achieved using an Extrap Soil DNA Kit Plus ver.2 (NIPPON STEEL Eco-Tech Co., Japan). The amount of 16S rRNA genes was measured by the real-time PCR method. The V4-V5 region of the 16S rRNA gene was PCR amplified using the primer set U515F (18mer, 5'-GTGYCAGCMGCCGCGGTA-3') and 926R (19mer, 5'-CCGYCAATTCMTT-RAGTT-3'). After purifying the obtained PCR amplification products, the double strand DNA concentration was measured

using the PicoGreen dsDNA Assay Kit (Invitrogen, USA). The concentration of PCR amplification products was sequenced with MiSeq system (Illumina, USA), and about 250 bases were analyzed from each side of PCR amplification products. The obtained array data were interpreted using the quantitative insights into the microbiological ecology (QIIME2) pipeline.

3. Results and discussion

3.1 Anaerobic batch experiment: Effects of additive concentration on CO₂ biomethanation

This experiment aimed to investigate the effect of different additives concentrations on biomethanation of gaseous CO₂. The gas change in the headspace according to the concentrations of the additives is shown in Figure 1.

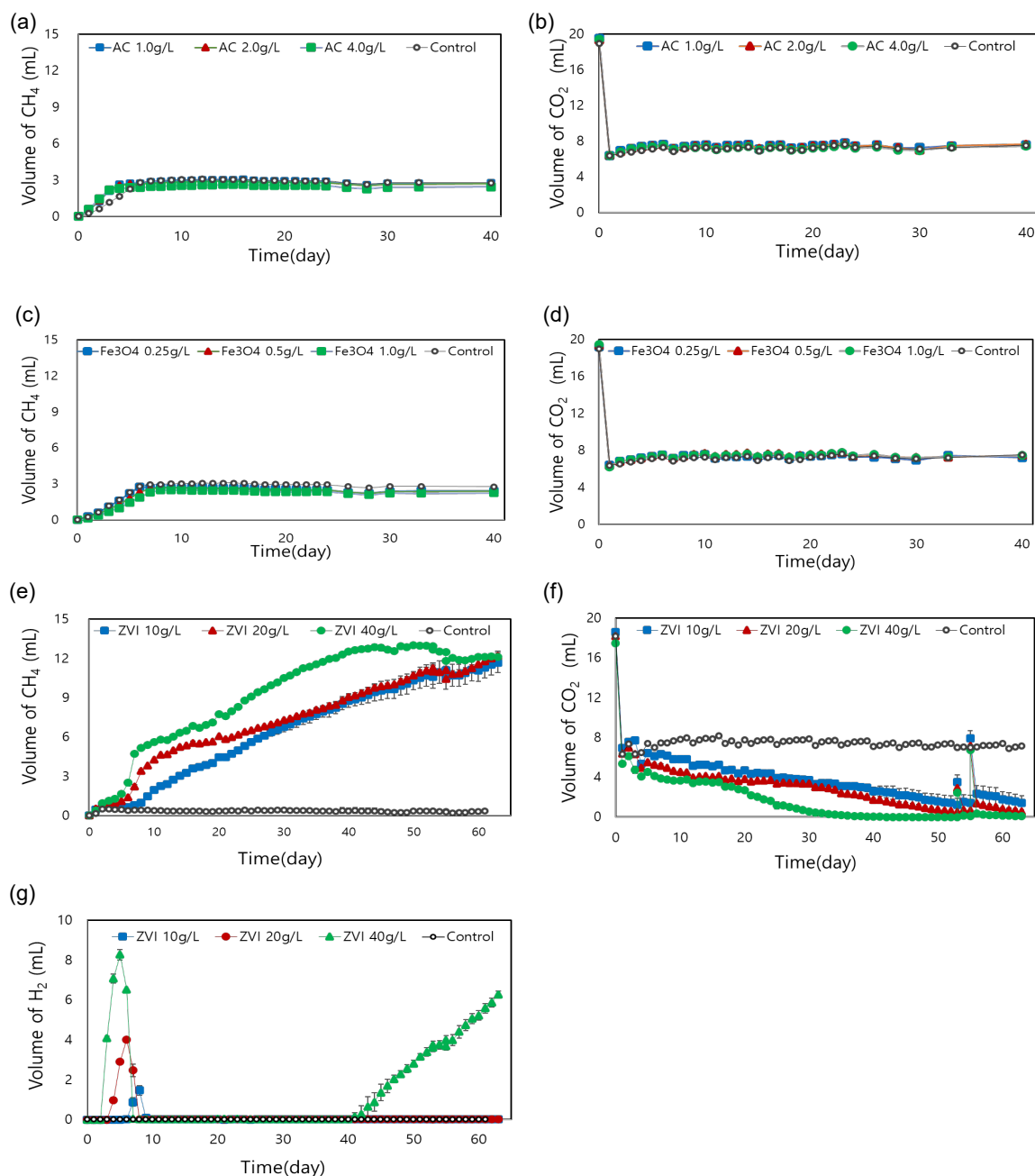


Figure 1: Gas volume of headspace with different additives concentrations: (a) CH₄ with the addition of AC, (c) Fe₃O₄ and (e) ZVI, (b) CO₂ with the addition of AC, (d) Fe₃O₄ and (f) ZVI, (g) H₂ with the addition of ZVI

CH₄ production in the case of AC decreased as the AC concentrations increased (Figure 1a). The CO₂ in the case of AC was dissolved in water as in all samples and decreased rapidly, but there was no significant change with a difference of ± 1.22 % from the control thereafter (Figure 1b). Fe₃O₄ had the similar tendency as CH₄ and CO₂ in the case of AC. These results suggested that CH₄ production was inhibited by all concentrations of AC and Fe₃O₄. In the case of AC, the time to reach the maximum CH₄ production was faster than control, and this acceleration indicates that the addition of AC does not inhibit microorganisms (Li et al., 2019). Fe₃O₄ can inhibit CH₄ production by using electrons necessary for CH₄ production (Zhao et al., 2018).

Figure 1e shows the CH₄ changes of the ZVI. CH₄ increased slowly for 6 d and increased sharply until 10 d, with the highest production rates of 0.69 and 0.93 mL/day at 20 and 40 gZVI/L due to H₂ generation. Since then, it has been investigated that 20 gZVI/L produced almost constant CH₄ production rate. 40 gZVI/L accelerated CH₄ production from the 22 d but gradually decreased due to decrease in CO₂. Due to the lack of CO₂ in 40 gZVI/L, 2 mL and 5 mL of CO₂ were injected into all ZVI samples on the 53 and 55 d, but CH₄ was not produced at 40 gZVI/L. CH₄ production at 10 gZVI/L began in earnest on the 8 d and was relatively constant at 0.19 mL/day until the end. Initially, CO₂ of the ZVI showed a similar rapid dissolution of CO₂ as AC and Fe₃O₄ (Figure 1f). After that, the CO₂ reduction rate had the opposite result according to the increase in CH₄, and the CO₂ increased on 53 and 55 d when CO₂ was additionally injected but decreased sharply again the next day. H₂ increased in proportion to the ZVI concentration, and all reacted within 10 d. After 10 d, H₂ has not been produced, but it has been produced at 40 gZVI/L since CO₂ became zero after 40 d. To confirm the ZVI surface corrosion according to the size of the ZVI, the ZVI was classified by size before and after the experiment (Figure 2). It decreased by 0.035 % in 70-100 μ m, 2.83 % in 50-70 μ m and 7.16 % in 32-50 μ m but increased by 10.025 % in 20-32 μ m. This indicates that the smaller the ZVI size, the faster the corrosion occurs, and the reason for H₂ generation after 40 d in (Figure 1g). It is assumed that CH₄ was not produced at 40 gZVI/L after 40 d due to ammonia production caused by H₂ and precipitation with Fe²⁺ dissolved in water before CH₄ production (Rocamora et al., 2023).

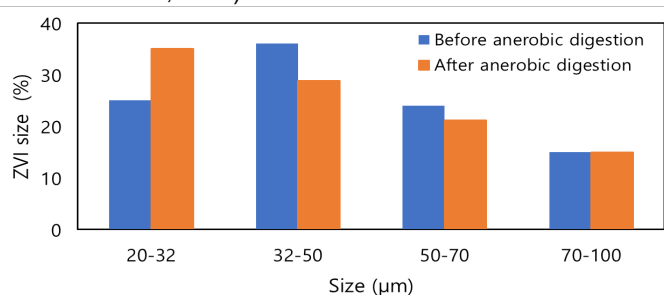


Figure 2: Changes in ZVI size before and after anaerobic digestion

3.2 Effects of additives concentration on the VFAs concentrations and pathway of CO₂

The TIC and VFAs concentration were analyzed at the finish of the experiment, and Table 1 displays the TIC and VFAs concentrations as well as CO₂ conversion rates. Only propionic acid and isobutyric acid were identified as a result of VFAs analysis. The pH of control, AC, and Fe₃O₄ was in a narrow range of 7.29-7.34, slightly increased from the initial 7.12. The pH of the ZVI was 7.68-7.77, which was increased by the reaction of H₂. The TIC and VFAs concentrations of the controls were analyzed at 7.39 mg/L and 35 mg/L. The higher the concentration of ZVI, the lower the concentration of TIC and VFAs, indicating that biomethanation progresses rapidly at high concentrations of ZVI. Considering that AC and Fe₃O₄ have higher VFAs concentration than control and ZVI at all concentrations, VFAs are accumulating. The accumulation of VFAs caused microbial inhibition in AC and Fe₃O₄, leading to the inhibition of CO₂ use and CH₄ production (Zhang et al., 2018). Additional AC addition is considered because a study investigated that adding more than 5 g/L of AC is effective in preventing VFAs inhibition (Johnravindar et al., 2020). Electron exchange by electron generated in AC and Fe₃O₄ induces CO₂ availability, resulting in higher TIC and VFAs concentrations than control (Vyrides et al., 2018).

CO₂ excluding CH₄, VFAs, and TIC was calculated as unidentified CO₂. Unidentified CO₂ can be caused by the production of by-products such as ethanol in control, precipitation of Fe²⁺ in ZVI and Fe₃O₄, and CO₂ adsorption in AC (Wu et al., 2021). Unidentified CO₂ increased with lower concentrations of CH₄ production, TIC and VFAs concentrations. In Control, CO₂ loss was the highest at 84.51 %, and in the case of ZVI, high CH₄ production reduced unidentified CO₂ to 51.21-52.68 %. ZVI was optimized with 4.8-8.1 % lower CO₂ to VFAs and 10 % higher CO₂ to CH₄ compared to a previous study (Dong et al., 2022). Subsequently, it had unidentified CO₂ of 71.58-77.37 % in AC with high TIC and VFAs concentrations, and Fe₃O₄ had unidentified CO₂ of 76.93-80.69 %, and

the highest except for control. Unidentified CO₂ was lower than the control in all conditions, suggesting that the use of additives induces the conversion of CO₂ to VFAs and inorganic carbon, which are necessary for biomethanation.

Table 1: TIC and VFAs concentration and CO₂ conversion ratio

	Control ZVI (g/L)			AC (g/L)		Fe ₃ O ₄ (g/L)				
	-	10	20	40	1	2	4	0.25	0.5	1
VFAs concentration (mg/L)	38.5±1	19.5±13	7.5±0.7	0	39.5±9	53.5±4	49.5±1	46.5±8	50.5±28	68.5±5
TIC (mg/L)	7.39	16.87	15.00	4.47	34.02	23.76	42.92	5.35	5.29	30.14
CO ₂ to CH ₄ (%)	0	43.72	44.62	47.91	0	0	0	0	0	0
CO ₂ to VFAs (%)	13.9	3.43	1.27	0	15.42	21.27	19.34	18.18	19.83	16.72
Dissolved CO ₂ (%)	1.59	1.64	1.43	0.43	7.21	5.11	9.08	1.13	1.12	6.35
Unidentified CO ₂ (%)	84.51	51.21	52.68	51.66	77.37	73.62	71.58	80.69	79.05	76.93

3.3 Microbial community structure analysis

PCR analysis was conducted to investigate the effect of ZVI addition on microbial change (Figure 3). Various groups of methanogens were found depending on the type, such as hydrogenotrophic methanogens (Methanobrevibacter, Methanobacterium, Methanoculleus, and Methanolinea), acetoclastic methanogens (Methanosaeta and Methanosarcina) and methyltrophic methanogens (Candidatus Methanofastidiosum and Methanomassiliicoccus) were found. The addition of ZVI had a positive effect on the diffusion of hydrogenotrophic and methyltrophic methanogens and acetoclastic methanogens decreased accordingly.

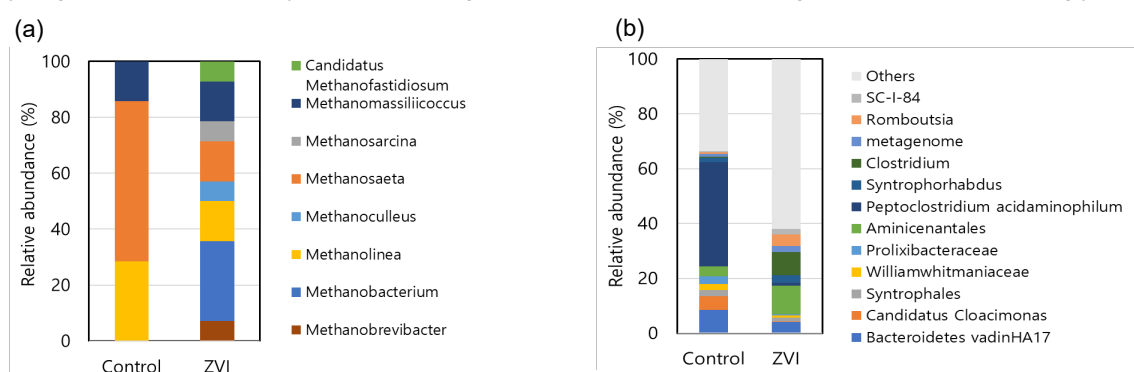


Figure 3: The microbial community in control and 40 gZVI/L, the amounts of microbial genus of (a) archaea and (b) bacteria

Control had a simpler classification than ZVI, dominated by acetoclastic methanogens, and on the contrary, ZVI produced a variety of archaea and changed its dominant structure from acetoclastic methanogens to hydrogenotrophic methanogens. In the control group, Methanosaeta was dominant, but since Methanosaeta can only utilize acetic acid, CH₄ was not produced in the control group where only propionic and isobutyric acid were present (Mori et al., 2012). In ZVI, Methanobacterium was dominant, and this can convert propionic acid into biogas (Han et al., 2020). In bacteria, Peptoclostridium acidaminophilum was dominant in control and clostridium and aminicenantales were dominant in ZVI. Clostridium is one of the typical genera contained in homoacetogens, which can utilize H₂ and CO₂ to produce VFAs (Dong et al., 2022). Aminicenantales is a functional bacterial product associated with hydrolysis and acidification and has the potential to decompose inhibitory compounds (Fan et al., 2022). Unlike previous studies where acetic acid was the predominant VFAs, the presence of propionic and isobutyric acids led to the proliferation of Methanobacterium rather than Methanolinea (Andronikou et al., 2022). Additionally, various archaea species were observed to grow instead of being limited to one or two species (Dong et al., 2022). The addition of ZVI enabled the degradation of various VFAs by growing various archaea, resulting in the biomethanation of CO₂.

4. Conclusion

This study centered on exploring the function of electron donors in the biochemical conversion of CO₂ to CH₄. This study reconfirmed the biomethanation of gaseous CO₂ through ZVI, but the addition of AC and Fe₃O₄ did

not lead to biomethane generation due to the accumulation of VFAs. The addition of AC and Fe₃O₄ increased the concentrations of VFAs and TIC, both of which are essential for biomethanation. This serves as evidence that the addition of AC and Fe₃O₄ supports the potential for biomethanation. The addition of ZVI facilitated the growth of various microbial species, resulting in the production of biomethane. The study found that ZVI exhibited varying H₂ generation rates depending on its size, and under conditions of CO₂ deficiency, inhibition occurred due to the H₂ produced by ZVI. Maintaining adequate CO₂ supply is essential to prevent this inhibition caused by H₂ generated from ZVI. While this study demonstrated the potential of electron donors for the biomethanation of CO₂, a significant portion of the CO₂ remains unidentified for, indicating the need for broader analyses and in-depth investigations to ascertain the pathways of CO₂.

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